Number 21, 1966 781

## Azoalkenes as Intermediates in the Formation of Osazones from Steroid α-Halogeno-ketones

By J. Buckingham and R. D. Guthrie (The Chemical Laboratory, University of Sussex, Brighton)

The recent Communication by Caglioti and his coworkers¹ on the formation of cyclohexan-1,2dione phenylosazone from 2-acetoxycyclohexanone, with 1-phenylazocyclohexene as intermediate, prompts us to report our findings in this field.

Following the observation that the so-called phenylhydrazone of  $3\beta$ ,  $5\alpha$ -dihydroxycholestan-6-one was in fact 6-phenylazocholest-5-en- $3\beta$ -ol<sup>4</sup> and that the latter compound could be prepared from other  $5\alpha$ - and  $5\beta$ -substituted cholestan-6-ones, we attempted to extend this elimination reaction to 2-substituted cholestan-3-ones and phenylhydrazine. The reaction of 2,4-dinitrophenylhydrazine with steroid  $\alpha$ -bromo-ketones has been previously studied. 5-7

Reaction of  $2\alpha$ -bromocholestan-3-one with phenylhydrazine in boiling ethanol containing acetic acid gave an almost immediate precipitate of the orange-yellow 3-phenylazocholest-2-ene (I)<sup>8</sup> m.p. 178—179°,  $[\alpha]_D + 94^\circ$ . This substance was distinct from the colourless phenylhydrazone (III) of cholest-1-en-3-one, m.p.  $205^\circ$   $[\alpha]_D + 89^\circ$ . This contrasts with 2,4-dinitrophenylhydrazine reaction<sup>5,6</sup> where the product formed from  $2\alpha$ -bromocholestan-3-one was the dinitrophenylhydrazone of cholest-1-en-3-one. The ene-azo-compound was not isomerised to the hydrazone (III) on treatment with either hot alcoholic acid or base.

Treatment of 2,2-dibromocholestan-3-one with

phenylhydrazine under the same conditions as above gave 2-bromo-3-phenylazocholest-2-ene (II) m.p.  $164-165^{\circ}$ ,  $[\alpha]_{\rm D}+103^{\circ}$ . Recrystallisation from acetone containing a trace of acid caused an immediate isomerisation to the almost colourless phenylhydrazone (IV) of 2-bromocholest-1-en-3-one m.p.  $180-182^{\circ}$  (dec.)  $[\alpha]_{\rm D}+57^{\circ}$  also obtained from phenylhydrazine and the parent ketone.

Further treatment of either compounds (I) or (II) with phenylhydrazine in ethanolic acetic acid gave the phenylosazone of cholestan-2,3-dione, (V) m.p. approx.  $206-209^{\circ}$  (dec.)  $[\alpha]_{\rm p}+132^{\circ}$ , in good yield. Under the conditions employed no intermediate phenylhydrazino-phenylhydrazone (cf., ref. 1) was isolable. Nevertheless this demonstrates that azo-enes are intermediates in the formation of osazones from  $\alpha$ -halogeno-ketones, and extends the findings of Caglioti.

The phenylhydrazone (IV) of 2-bromocholest-1en-3-one was also obtained from the reaction of phenylhydrazine with  $1\alpha, 2\beta$ -dibromocholestan-3one, implying that elimination from this compound occurs from the ene-hydrazine form of the initiallyformed phenylhydrazone. Similar  $\alpha\beta$ -unsaturated hydrazones have been isolated from 2-deoxy-sugar derivatives:9,10 their formation again presumably occurring via the ene-hydrazine form of the initially formed hydrazones. Another reaction relevant to the present study is that of phenylhydrazine with aldehydo-D-glucose penta-acetate to give D-arabino-3,4,5,6-tetra-acetoxy-1-phenylazo-trans-hex-1-ene.1

(Received, September 29th, 1966; Com. 725.)

L. Caglioti, G. Rosini, and F. Rossi, J. Amer. Chem. Soc., 1966, 88, 3865.
 R. H. Pickard and J. Yates, J. Chem. Soc., 1908, 1678.
 L. F. Fieser and S. Rajagopalan, J. Amer. Chem. Soc., 1949, 71, 3938.
 J. Buckingham and R. D. Guthrie, unpublished work.

<sup>5</sup> C. Djerassi, J. Amer. Chem. Soc., 1949, 71, 1003.
<sup>6</sup> V. R. Mattox and E. C. Kendall, J. Amer. Chem. Soc., 1948, 70, 882; 1950, 72, 2290.
<sup>7</sup> The reaction of arylhydrazines with other α-halogeno-carbonyl compounds has been reported in the literature. F. D. Chattaway and H. Irving, J. Chem. Soc., 1930, 88; J. van Alphen, Rec. Trav. chim., 1945, 64, 109.

8 All new compounds were crystalline and gave satisfactory analyses and spectra. All rotations are for chloroform

<sup>9</sup> M. G. Blair, D. Lipkin, J. C. Sowden, and D. R. Strobach, J. Org. Chem., 1960, 25, 1679.

<sup>10</sup> M. Miyamoto, Y. Kawamatsu, M. Shinohara, Y. Nakadaira, and K. Nakanishi, Tetrahedron, 1966, 22, 2785.

<sup>11</sup> M. L. Wolfrom, A. Thompson, and D. R. Lineback, J. Org. Chem., 1962, 27, 2563.